

## PhI=NSes: A New Iminoiodinane Reagent for the Copper-Catalyzed Aziridination of Olefins

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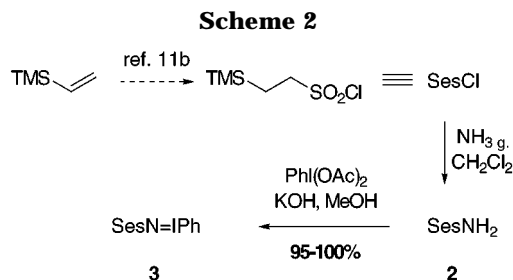
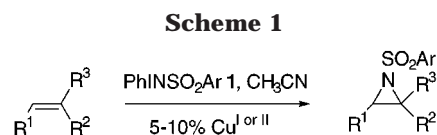
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### Introduction

In addition to their incorporation in the structures of natural and/or biologically active products, aziridines are of paramount importance in organic synthesis since they are valuable precursors of amino sugars, alkaloids, or substituted  $\alpha$ -amino acids.<sup>1</sup> Among the numerous routes to this heterocycle, the copper-catalyzed aziridination of olefins developed by Evans<sup>2</sup> has become a method of choice, allowing the formation of aziridines directly by reaction with a nitrene generated from an iminophenyl iodinane (Scheme 1). The value of this process has recently been demonstrated in the total synthesis of natural products and dipeptide isosteres.<sup>3</sup>

So far, only [*N*-(arenesulfonyl)imino]phenyliodines **1** have been used as nitrene sources, leading to *N*-(arenesulfonyl)aziridines.<sup>4</sup> Although several methods of deprotection of sulfonamides have been reported in the literature,<sup>5</sup> their cleavage sometimes proves to be troublesome.<sup>3a,6</sup> Thus, it appeared to us that preparation of *N*-[2-(trimethylsilyl)ethanesulfonyl]aziridines (*N*-(Ses)-aziridines) would provide an interesting alternative to the (arenesulfonyl)aziridines in terms of protection–deprotection strategy.

Until now, no aziridinations using [*N*-(alkylsulfonyl)imino]phenyliodines have been described, probably because the *N*-methylsulfonyl derivative has been claimed to be unstable and explosive.<sup>7</sup> To the best of our knowledge, a single report deals with the attempt to generate a nitrene species from this ylide.<sup>8</sup> Despite this, preliminary experiments with this iodine have allowed us to



prepare in rather modest yields *N*-(methylsulfonyl)aziridines from olefins<sup>9</sup> but their deprotection would require harsh conditions incompatible with many functional groups.<sup>10</sup>

On the basis of this result, we next turned our attention to the Ses group. This protecting group, originally developed by Weinreb,<sup>11</sup> has found numerous applications in total synthesis.<sup>12</sup> More significantly, *N*-(Ses)aziridines<sup>13a</sup> have been successfully used for the preparation of substituted  $\alpha$ -amino acids<sup>13b</sup> and oligosaccharides.<sup>13c</sup> In this context, we report herein the preparation and the synthetic application of the [*N*-(trimethylsilyl)ethanesulfonyl]imino]phenyliodine (*PhI=NSes*) **3**, the first [*N*-(alkylsulfonyl)imino]phenyliodine, for the copper-catalyzed aziridination of olefins leading to synthetically versatile *Ses*-protected aziridines.

### Results and Discussion

Following the protocol described for the isolation of the iodines  $PhI=NSO_2Ar$ ,<sup>4,14</sup> the reaction of sulfonamide **2** (prepared from the corresponding sulfonyl chloride according to the recently published procedure)<sup>11b</sup> with iodosobenzene diacetate in the presence of potassium hydroxide gave the reagent **3** (Scheme 2). However, a slight modification of the workup was needed because the nitrene precursor **3** could not be crystallized from the medium. Thus, despite the sensitivity of the phenyliodines to water, dilution of the reaction mixture with dichloromethane, followed by an aqueous wash with ice–water, allowed the isolation of  $PhI=NSes$  **3** in quantitative yield as a stable yellow solid of analytical purity, which can be stored at  $-20\text{ }^\circ\text{C}$  under argon.

In the presence of a catalytic amount of  $Cu(OTf)_2$ , a slight excess of the iodine **3** reacted with olefins **4** (the

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**Table 1. Copper-Catalyzed Aziridination of Representative Olefins**

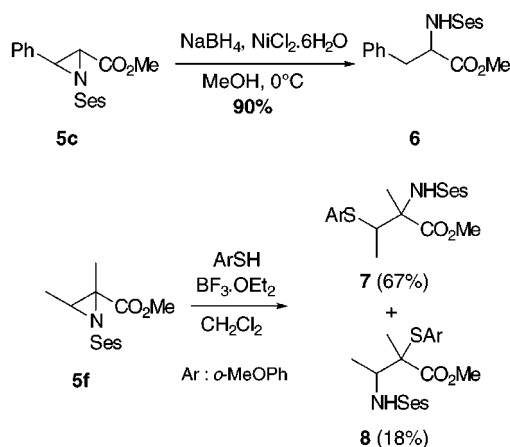
Entry	Substrate	Catalyst	% Yield <sup>a</sup>
1		Cu(OTf) <sub>2</sub> CuOTf	5a : 58 5a : 68
2		Cu(OTf) <sub>2</sub>	5b : 40
3		Cu(OTf) <sub>2</sub> CuOTf	5c : 37 (95) <sup>b,c</sup> 5c : 39 (85) <sup>b</sup>
4		Cu(OTf) <sub>2</sub>	5d : 49 <sup>d</sup>
5		Cu(OTf) <sub>2</sub> CuOTf	5e : 52 <sup>e</sup> 5e : 60
6		Cu(OTf) <sub>2</sub>	5f : 47 <sup>e</sup>
7		Cu(OTf) <sub>2</sub>	5g : 48 (75) <sup>b</sup>
8		Cu(OTf) <sub>2</sub>	5h : 34
9		Cu(OTf) <sub>2</sub> CuOTf	5i : 55 5i : 67
10		Cu(OTf) <sub>2</sub> CuOTf	5j : 33 5j : 43

<sup>a</sup> Isolated yield after flash chromatography. <sup>b</sup> Value in parentheses for yield based on consumed starting olefin. <sup>c</sup> 2 equiv of olefin was used. <sup>d</sup> 5 equiv of olefin was used. <sup>e</sup> 63% (entry 5) and 53% (entry 6) of aziridines were, respectively, isolated when 1.7 equiv of PhI=NSes was used.

stoichiometrically limiting component except in entries 3 and 4, Table 1) to form the *N*-(Ses)aziridines **5**. The yields were in the 35–60% range and are comparable to those obtained with PhI=NTs.<sup>2,4,15</sup> Moreover, the yields could be substantially increased by the use of the copper(I) salt, CuOTf, instead of Cu(OTf)<sub>2</sub> (Table 1, entries 1, 3, 5, 9, and 10), or by adding a larger quantity of the nitrene precursor (Table 1, entries 5 and 6). The reaction takes place with terminal, electron-rich, or electron-poor olefins. It proceeds in a stereospecific manner with *trans*-olefins since only *trans*-aziridines derived from methyl cinnamate and methyl tiglate were isolated (Table 1, entries 3 and 6). In the case of norbornene, as noted by Evans,<sup>2</sup> the *exo*-aziridine was the sole isomer produced. Finally, as recently described by us for nitrene precursors of type **1**, the 2-substituted acrylates show higher reactivity than methyl acrylate itself (Table 1, entries 5–7 vs entry 4).<sup>15</sup>

As previously demonstrated by Wipf,<sup>13b</sup> *N*-(Ses)aziridines are sufficiently activated to allow ring opening by nucleophiles. Thus, the 3-phenylaziridine derivative **5c** reacted with sodium borohydride/nickel chloride to give exclusively the *N*-Ses-protected phenylalanine **6** in 90% yield, while aziridine **5f** reacted with a soft nucleophile such as *o*-methoxythiophenol under mild conditions to give mainly the C-3 aziridine ring-opened product **7** together with a minor quantity of the product of C-2 attack **8** (Scheme 3).

From a synthetic point of view, the *N*-(Ses) protecting group offers the opportunity of mild deprotection of the

**Scheme 3****Table 2. Deprotection of the *N*-Ses Derivatives**

Compound	Substrate (R = Ses)	Conditions <sup>a</sup>	% Yield (R = H) <sup>b</sup>
7		A	9 : 95
5a		B	10 : 62
5b		C	11 : 60
5i		C	12 : 60
5j		C	13 : 68

<sup>a</sup> Key: (A) 3 equiv of CsF, DMF, 90 °C; (B) 4 equiv of TASF, DMF, rt; (C) 4 equiv of TASF, acetonitrile, rt. <sup>b</sup> Isolated yield after flash chromatography.

resulting sulfonamides by use of cesium fluoride in DMF.<sup>11</sup> Thus, under these conditions, compound **6** could be efficiently transformed into the amino derivative **9** in 95% yield (Table 2).

More significantly, the *N*-(Ses)aziridines themselves could be deprotected without provoking opening of the aziridine ring. Except for a few isolated examples, no general procedure exists for this type of transformation.<sup>5c,16,17</sup> While the above conditions (CsF in DMF) applied to aziridines **5** unexpectedly failed to give the corresponding deprotected three-membered ring, it was found that use at room temperature of tris(dimethylamino)sulfonium difluorotrimethylsilicate (TASF), a reagent recently found by Roush to cleave silicon protecting groups under mild conditions,<sup>18</sup> provided the NH aziridine products **10**, **11**, **12**, and **13**, respectively, in good yields (nonoptimized, Table 2). To the best of our knowledge, TASF has never been employed before to cleave the Ses protecting group.<sup>19</sup> However, we observed that this procedure failed with the acrylate-derived aziridines **5d–f**. Nevertheless, it should be pointed out that the conditions developed by Andersson<sup>17</sup> for deprotection of cyclo-

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hexyl *N*-(tosyl)aziridine-2-carboxylates (i.e., reduction with magnesium in methanol) were recently found to give aziridine ring-opened product when applied to methyl aziridine-2-carboxylates.<sup>20</sup>

In conclusion,  $\text{PhI}=\text{NSes}$  **3** represents the first [(*N*-(alkylsulfonyl)imino)phenyliodinane useful for the copper-catalyzed aziridination of olefins. In comparison to  $\text{PhI}=\text{NTs}$ , its isolation is much easier while their reactivities are comparable. Moreover, this new iodine allows the formation of *N*-(*Ses*)aziridines that can, in turn, be opened by nucleophiles under mild conditions and/or deprotected at the nitrogen position without side reactions. In the latter case, use of the hypervalent silicon reagent TASF for deprotection of the *Ses* group is very convenient. These *N*-(*Ses*)aziridines are thus of high synthetic interest. Extension of this work to the asymmetric copper-catalyzed aziridination of olefins is currently under investigation.

### Experimental Section

**General Methods.** Melting points are uncorrected. IR spectra of samples were obtained as films (i.e., by application of a  $\text{CHCl}_3$  solution to a NaCl plate followed by evaporation of the solvent).  $^1\text{H}$  and  $^{13}\text{C}$  NMR chemical shifts are given as  $\delta$  values with reference to  $\text{Me}_4\text{Si}$  as internal standard. Thin-layer chromatography was performed on Merck silica gel 60 plates with a fluorescent indicator. The plates were visualized with UV light (254 nm) and with a 3.5% solution of phosphomolybdic acid in ethanol. All column chromatography was conducted on Merck 60 silica gel (230–240 mesh) at medium pressure (200 mbar). All solvents were distilled and stored over 4 Å molecular sieves before use. Elemental analyses were performed at the ICSN, CNRS, Gif-sur-Yvette, France.

**[*N*-(2-(Trimethylsilyl)ethanesulfonyl)imino]phenyliodinane (3).** To a stirred solution of 2-(trimethylsilyl)ethanesulfonamide (**2**, 1.19 g, 6.56 mmol, 1.00 equiv) and potassium hydroxide pellets (0.92 g, 16.4 mmol, 2.50 equiv) in dry methanol (20 mL) held at 0 °C under argon was added iodobenzene diacetate (2.12 g, 6.56 mmol, 1.00 equiv). The reaction mixture was kept at 0 °C for 30 min and then allowed to warm to 20 °C. After being stirred for 3 h, the homogeneous yellow solution was cooled before being diluted with dichloromethane (30 mL) and washed with ice-water (20 mL). The organic phase was dried with  $\text{MgSO}_4$  and then evaporated to dryness to give a pale yellow solid (2.50 g, 6.52 mmol, 99%): mp 84–85.5 °C; IR (film) 2953, 1441, 1330, 1245, 1225, 1165, 1145, 1115, 885, 860, 735  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  -0.06 (s, 9H), 0.92 (m, 2H), 2.84 (m, 2H), 7.40 (t, 2H,  $J = 7.7$  Hz), 7.52 (d, 1H,  $J = 7.4$  Hz), 7.96 (d, 2H,  $J = 7.9$  Hz). Anal. Calcd for  $\text{C}_{11}\text{H}_{18}\text{INO}_2\text{SSi}$ : C, 34.47; H, 4.73; N, 3.65; S, 8.36. Found: C, 34.61; H, 4.74; N, 3.75; S, 8.51.

**Typical Aziridination Procedure (unless Otherwise Noted).**  $\text{PhI}=\text{NSes}$  **3** (1.3 equiv) was added portionwise, over a period of 3 h, to a mixture held under argon of 4 Å molecular sieves (~150 mg),  $\text{Cu}(\text{OTf})_2$  (10 mol %), and olefin (1 equiv) in acetonitrile (1.6 mL). The green reaction mixture was stirred at room temperature for 24 h and then purified directly by flash chromatography on silica gel, affording aziridines **5**.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-phenylaziridine (5a).** Starting from 69  $\mu\text{L}$  (0.6 mmol) of styrene, 0.100 g (0.35 mmol, 58%) of aziridine was isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 8:1):  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.03 (s, 9H), 1.14 (m, 2H), 2.42 (d, 1H,  $J = 4.4$  Hz), 2.98 (d, 1H,  $J = 7.1$  Hz), 3.13 (m, 2H), 3.71 (dd, 1H,  $J = 4.4, 7.1$  Hz), 7.34 (m, 5H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$  -2.1, 9.7, 35.1, 40.4, 49.1, 126.5, 128.4, 128.7, 135.1; mass spectrum (CI)  $m/z$  284 ( $\text{M} + \text{H}^+$ ). Anal. Calcd for  $\text{C}_{13}\text{H}_{21}\text{NO}_2$ -

$\text{SSi}$ : C, 55.08; H, 7.47; N, 4.94; S, 11.31. Found: C, 54.87; H, 7.27; N, 4.96; S, 11.36.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-(2-phenylethyl)aziridine (5b).** Starting from 150  $\mu\text{L}$  (1 mmol) of 4-phenyl-1-butene, 0.125 g (0.40 mmol, 40%) of aziridine was isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 6:1):  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.06 (s, 9H), 1.13 (m, 2H), 1.86 (m, 2H), 2.06 (d, 1H,  $J = 4.6$  Hz), 2.57 (d, 1H,  $J = 7$  Hz), 2.78 (m, 3H), 3.05 (m, 2H), 7.19–7.29 (m, 5H);  $^{13}\text{C}$  NMR (62.5 MHz,  $\text{CDCl}_3$ )  $\delta$  -1.9, 9.8, 33.1, 33.4, 33.8, 38.3, 48.9, 126.3, 128.4, 128.7; mass spectrum (CI)  $m/z$  312 ( $\text{M} + \text{H}^+$ ). Anal. Calcd for  $\text{C}_{15}\text{H}_{25}\text{NO}_2\text{SSi} \cdot 0.25\text{H}_2\text{O}$ : C, 57.01; H, 8.13; N, 4.43; S, 10.14. Found: C, 57.03; H, 7.97; N, 4.29; S, 10.21.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-(methoxycarbonyl)-3-phenylaziridine (5c).** Starting from 146 mg (0.9 mmol) of *trans*-methyl cinnamate and in the presence of 50 mg of  $(\text{CuOTf})_2 \cdot \text{C}_6\text{H}_6$ , 120 mg (0.35 mmol, 39%) of aziridine together with 65 mg (0.39 mmol, 44%) of unreacted cinnamate were isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 9:1 then 8:1):  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.04 (s, 9H), 1.12 (m, 2H), 3.18 (m, 2H), 3.45 (d, 1H,  $J = 3.8$  Hz), 3.87 (s, 3H), 4.40 (d, 1H,  $J = 3.8$  Hz), 7.36 (m, 5H);  $^{13}\text{C}$  NMR (62.5 MHz,  $\text{CDCl}_3$ )  $\delta$  -2.0, 9.9, 47.0, 47.2, 52.0, 53.2, 127.0, 128.8, 129.1, 133.3, 166.3; mass spectrum (CI)  $m/z$  342 ( $\text{M} + \text{H}^+$ ). Anal. Calcd for  $\text{C}_{15}\text{H}_{23}\text{NO}_4\text{SSi} \cdot 0.25\text{H}_2\text{O}$ : C, 52.07; H, 6.85; N, 4.05. Found: C, 51.73; H, 6.66; N, 4.22.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-(methoxycarbonyl)aziridine (5d).** Starting from 360  $\mu\text{L}$  (4.0 mmol) of methyl acrylate dissolved in acetonitrile (2 mL) and 310 mg (0.8 mmol) of  $\text{PhI}=\text{NSes}$ , 0.105 g (0.395 mmol, 49%) of aziridine was isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 4:1):  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.07 (s, 9H), 1.13 (m, 2H), 2.60 (d, 1H,  $J = 4.1$  Hz), 2.75 (d, 1H,  $J = 7.0$  Hz), 3.16 (m, 2H), 3.30 (dd, 1H,  $J = 4.1, 7.0$  Hz), 3.80 (s, 3H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$  -2.0, 9.5, 31.6, 34.8, 49.3, 53.0, 167.5; mass spectrum (CI)  $m/z$  266 ( $\text{M} + \text{H}^+$ ). Anal. Calcd for  $\text{C}_9\text{H}_{19}\text{NO}_4\text{SSi}$ : C, 40.73; H, 7.22; N, 5.28; S, 12.08. Found: C, 40.81; H, 7.04; N, 5.21; S, 11.89.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-(methoxycarbonyl)-2-methylaziridine (5e).** Starting from 128  $\mu\text{L}$  (1.2 mmol) of methyl methacrylate, 0.175 g (0.63 mmol, 52%) of aziridine was isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 4:1):  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.07 (s, 9H), 1.12 (m, 2H), 1.84 (s, 3H), 2.65 (s, 1H), 2.86 (s, 1H), 3.14 (m, 2H), 3.77 (s, 3H);  $^{13}\text{C}$  NMR (62.5 MHz,  $\text{CDCl}_3$ )  $\delta$  -1.9, 9.7, 15.7, 39.2, 45.2, 51.5, 53.1, 169.1; mass spectrum (CI)  $m/z$  280 ( $\text{M} + \text{H}^+$ ). Anal. Calcd for  $\text{C}_{10}\text{H}_{21}\text{NO}_4\text{SSi}$ : C, 42.98; H, 7.58; N, 5.01; S, 11.47. Found: C, 42.93; H, 7.43; N, 4.87; S, 11.51.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-(methoxycarbonyl)-2,3-dimethylaziridine (5f).** Starting from 96  $\mu\text{L}$  (0.8 mmol) of methyl tiglate, 0.110 g (0.37 mmol, 47%) of aziridine was isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 5:1):  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.04 (s, 9H), 1.08 (m, 2H), 1.31 (d, 3H,  $J = 5.8$  Hz), 1.48 (s, 3H), 3.08 (m, 2H), 3.63 (q, 1H,  $J = 5.8$  Hz), 3.75 (s, 3H);  $^{13}\text{C}$  NMR (62.5 MHz,  $\text{CDCl}_3$ )  $\delta$  -2.0, 9.7, 12.8, 15.3, 43.6, 51.1, 53.0, 168.8; mass spectrum (CI)  $m/z$  294 ( $\text{M} + \text{H}^+$ ). Anal. Calcd for  $\text{C}_{11}\text{H}_{23}\text{NO}_4\text{SSi} \cdot 0.03\text{C}_7\text{H}_{16}$ : C, 45.42; H, 7.98; N, 4.72; S, 10.81. Found: C, 45.46; H, 7.71; N, 4.67; S, 11.02.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-(*tert*-butoxycarbonyl)-2-phenylaziridine (5g).** Starting from 184 mg (0.9 mmol) of *tert*-butyl atropate, 165 mg (0.43 mmol, 48%) of aziridine together with 50 mg (0.24 mmol, 27%) of unreacted starting material were isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 8:1):  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.07 (s, 9H), 1.23 (m, 2H), 1.46 (s, 9H), 2.56 (s, 1H), 3.28 (m, 2H), 3.43 (s, 3H), 7.35–7.44 (m, 5H);  $^{13}\text{C}$  NMR (62.5 MHz,  $\text{CDCl}_3$ )  $\delta$  -2.0, 9.7, 27.6, 37.6, 50.1, 53.8, 83.6, 127.2, 128.5, 128.7, 135.1, 165.2; mass spectrum (CI)  $m/z$  384 ( $\text{M} + \text{H}^+$ ). Anal. Calcd for  $\text{C}_{18}\text{H}_{29}\text{NO}_5\text{SSi}$ : C, 56.36; H, 7.62; N, 3.65; S, 8.36. Found: C, 56.58; H, 7.62; N, 3.63; S, 8.32.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-7-azabicyclo[4.1.0]heptane (5h).** Starting from 81  $\mu\text{L}$  (0.8 mmol) of cyclohexene, 72 mg (0.275 mmol, 34%) of aziridine was isolated as a white solid after flash chromatography on silica gel (heptane–ethyl acetate 6:1): mp 57–58 °C;  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.06

(19) This could be an interesting alternative to the classical reagent (i.e.,  $\text{CsF}$  in DMF) since cases have been reported in which the *N*-(*Ses*) protecting group could not be removed under the standard conditions. For example, see: Ward, D. E.; Gai, Y.; Kaller, B. F. *J. Org. Chem.* **1996**, *61*, 5498.

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(s, 9H), 1.08–1.15 (m, 2H), 1.23–1.35 (m, 2H), 1.39–1.51 (m, 2H), 1.89 (m, 4H), 2.95 (m, 2H), 3.00–3.07 (m, 2H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$  -2.0, 9.8, 19.5, 23.0, 39.1, 49.0; mass spectrum (CI)  $m/z$  262 ( $\text{M} + \text{H}$ ) $^+$ . Anal. Calcd for  $\text{C}_{11}\text{H}_{23}\text{NO}_2\text{SSi}$ : C, 50.53; H, 8.87; N, 5.36; S, 12.26. Found: C, 50.62; H, 8.61; N, 5.15; S, 12.26.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-3-azatricyclo[3.2.1.0<sup>2,4</sup>exo]octane (5i)**. Starting from 94 mg (1.0 mmol) of norbornene, 0.150 g (0.55 mmol, 55%) of aziridine was isolated as a white solid after flash chromatography on silica gel (heptane–ethyl acetate 5:1): mp 59.5–61 °C;  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.06 (s, 9H), 0.81 (d, 1H,  $J = 10$  Hz), 1.06–1.14 (m, 2H), 1.26–1.30 (m, 2H), 1.47–1.54 (m, 3H), 2.52 (s, 2H), 2.90 (s, 2H), 2.99–3.06 (m, 2H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  -2.0, 9.8, 25.7, 28.3, 35.9, 41.0, 49.2; mass spectrum (CI)  $m/z$  274 ( $\text{M} + \text{H}$ ) $^+$ . Anal. Calcd for  $\text{C}_{12}\text{H}_{23}\text{NO}_2\text{SSi}$ : C, 52.71; H, 8.48; N, 5.12; S, 11.72. Found: C, 52.43; H, 8.16; N, 5.02; S, 11.65.

***N*-(2-(Trimethylsilyl)ethanesulfonyl)-2-nonylaziridine (5j)**. Starting from 205  $\mu\text{L}$  (1.0 mmol) of 1-undecene, 0.110 g (0.33 mmol, 33%) of aziridine was isolated as a colorless oil after flash chromatography on silica gel (heptane–ethyl acetate 15:1):  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.07 (s, 9H), 0.88 (t, 3H,  $J = 6.9$  Hz), 1.11–1.19 (m, 2H), 1.23–1.57 (m, 16H), 2.08 (d, 1H,  $J = 4.6$  Hz), 2.59 (d, 1H,  $J = 7.0$  Hz), 2.70 (m, 1H), 3.03–3.10 (m, 2H);  $^{13}\text{C}$  NMR (62.5 MHz,  $\text{CDCl}_3$ )  $\delta$  -1.9, 9.8, 14.1, 22.7, 26.9, 29.3, 29.4, 29.6, 31.6, 32.0, 33.4, 39.3, 48.9; mass spectrum (EI)  $m/z$  333 ( $\text{M}$ ) $^+$ . Anal. Calcd for  $\text{C}_{16}\text{H}_{35}\text{NO}_2\text{SSi}$ : C, 57.61; H, 10.57; N, 4.20; S, 9.61. Found: C, 57.38; H, 10.48; N, 4.11; S, 9.76.

**Reductive Ring Opening of the *N*-(Ses)aziridine 5c**. A solution of the *N*-(Ses)aziridine **5c** (80 mg, 0.234 mmol) in methanol (4 mL) was treated at 0 °C under argon with nickel chloride hexahydrate (55 mg, 0.234 mmol), and after the mixture had stirred for 5 min, sodium borohydride (88 mg, 2.34 mmol) was added in small portions. The reaction mixture was stirred for 1 h at 0 °C before saturated aqueous  $\text{NH}_4\text{Cl}$  (5 mL) was added. The mixture was extracted with dichloromethane (3  $\times$  10 mL), the combined organic phases were dried over  $\text{MgSO}_4$ , and the solvents were evaporated under vacuum. The residue was purified by column chromatography on silica gel (ethyl acetate–heptane 1:3), providing *N*-(Ses)-phenylalanine methyl ester **6** (72 mg, 80%) as a colorless oil:  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.07 (s, 9H), 0.92 (m, 2H), 2.67–2.84 (m, 2H), 3.08 (dd, 1H,  $J = 7.9, 13.8$  Hz), 3.26 (dd, 1H,  $J = 5.1, 13.8$  Hz), 3.88 (s, 3H), 4.44 (ddd, 1H,  $J = 9.4$  Hz), 7.30–7.42 (m, 5H); mass spectrum (CI)  $m/z$  344 ( $\text{M} + \text{H}$ ) $^+$ .

**Acid-Catalyzed Ring Opening of the *N*-(Ses)aziridine 5f**. To a solution of the *N*-(Ses)aziridine **5f** (76 mg, 0.260 mmol) and 2-methoxythiophenol (95  $\mu\text{L}$ , 0.780 mmol) in  $\text{CH}_2\text{Cl}_2$  (1.2 mL) was added at -20 °C under argon boron trifluoride etherate (64  $\mu\text{L}$ , 0.520 mmol). The reaction mixture was then allowed to come to 0 °C and stirred for 2 h before saturated aqueous  $\text{NaHCO}_3$  (2 mL) was added. The mixture was extracted with ethyl acetate (2  $\times$  5 mL), the combined organic phases were dried over  $\text{MgSO}_4$ , and the solvents were evaporated under vacuum. Elution of the chromatography column with ethyl acetate–heptane (1:5) first

gave compound **7** (75 mg, 67%) as a colorless oil;  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.04 (s, 9H), 1.04–1.12 (m, 2H), 1.33 (d, 3H,  $J = 7.1$  Hz), 1.72 (s, 3H), 2.95–3.08 (m, 2H), 3.32 (q, 1H,  $J = 7.1$  Hz), 3.72 (s, 3H), 3.92 (s, 3H), 5.73 (s, 1H), 6.92 (2d, 2H,  $J = 7.9$  Hz), 7.33 (dd, 1H), 7.45 (dd, 1H); mass spectrum (CI)  $m/z$  434 ( $\text{M} + \text{H}$ ) $^+$ .

Continued elution of the column with ethyl acetate–heptane (1:4) gave compound **8** (20 mg, 18%) as a colorless oil:  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.01 (s, 9H), 1.04–1.15 (m, 2H), 1.22 (s, 3H), 1.38 (d, 3H,  $J = 6.6$  Hz), 2.98–3.20 (m, 2H), 3.73 (s, 3H), 3.78 (m, 1H), 3.84 (s, 3H), 5.25 (d, 1H,  $J = 8.9$  Hz), 6.92 (m, 2H), 7.31–7.48 (m, 2H); mass spectrum (CI)  $m/z$  434 ( $\text{M} + \text{H}$ ) $^+$ .

**Typical Procedure for the Deprotection of the *N*-(Ses)-aziridines. Method B: 2-Phenylaziridine (10)**. A solution of the *N*-(Ses)aziridine **5a** (58 mg, 0.205 mmol) and TASF (225 mg, 4 equiv) in DMF (0.7 mL) was stirred at room temperature. After completion of the reaction as indicated by TLC, the mixture was purified by flash chromatography on silica gel (heptane–ethyl acetate 1:2), affording the deprotected aziridine **10** (15 mg, 0.126 mmol, 62%) as a yellow oil whose spectroscopic data were identical to those previously reported: $^{17}$   $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.15 (br s, 1H), 1.80 (d, 1H,  $J = 3.4$  Hz), 2.21 (d, 1H,  $J = 6.0$  Hz), 3.02 (dd, 1H,  $J = 3.4$  and 6.0 Hz), 7.22–7.34 (m, 5H); mass spectrum (CI)  $m/z$  120 ( $\text{M} + \text{H}$ ) $^+$ .

**Method C: 3-Azatricyclo[3.2.1.0<sup>2,4</sup>exo]octane (12)**. A solution of the *N*-(Ses)aziridine **5i** (74 mg, 0.270 mmol) and TASF (230 mg, 3 equiv) in acetonitrile (1 mL) was stirred at room temperature. After completion of the reaction as indicated by TLC, the mixture was purified by flash chromatography on silica gel (dichloromethane–ethanol 9:1), affording the deprotected aziridine **12** (18 mg, 0.164 mmol, 60%) as a colorless oil whose spectroscopic data were identical to those previously reported: $^{21}$   $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.70 (d, 1H,  $J = 10.7$  Hz), 0.99 (br s, 1H), 1.11–1.17 (m, 1H), 1.23–1.29 (m, 2H), 1.46–1.50 (m, 2H), 1.98 (s, 2H), 2.34 (s, 2H); mass spectrum (CI)  $m/z$  110 ( $\text{M} + \text{H}$ ) $^+$ .

**2-Phenylethylaziridine (11)**. Starting from 80 mg (0.257 mmol) of *N*-(Ses)aziridine **5b**, 23 mg (0.156 mmol, 60%) of the deprotected aziridine **11** was isolated as a yellow oil after flash chromatography on silica gel (dichloromethane–ethanol 9:1):  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.25 (br s, 1H), 1.35 (d, 1H,  $J = 3.5$  Hz), 1.69–1.78 (m, 3H), 1.94 (m, 1H), 2.77 (m, 2H), 7.17–7.28 (m, 5H); mass spectrum (CI)  $m/z$  148 ( $\text{M} + \text{H}$ ) $^+$ .

**2-Nonylaziridine (13)**. Starting from 110 mg (0.33 mmol) of *N*-(Ses)aziridine **5j**, 38 mg (0.224 mmol, 68%) of the deprotected aziridine **13** was isolated as a yellow oil after flash chromatography on silica gel (dichloromethane–ethanol 9:1):  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  0.88 (t, 3H,  $J = 6.9$  Hz), 1.19–1.50 (m, 18H), 1.77 (d, 1H), 1.95 (m, 1H); mass spectrum (CI)  $m/z$  170 ( $\text{M} + \text{H}$ ) $^+$ .

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